

## Writing Nuclear Equations Answer Key

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Nuclear Equations Answer Key Writing Nuclear Equations KEY Write nuclear equations that describe the following processes. 1. Uranium-235 undergoes an alpha decay to produce thorium-231. 2. Lanthanum -144 becomes cerium-144 when it undergoes a beta decay. 3. Neptunium-233 is formed when americium-237 undergoes a nuclear decay process. 4.

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Write the nuclear equation for this reaction and identify the radionuclide.  $\text{Pu } 242\text{Cm } 96 \text{ } 1 \text{ } 0 \text{ } 4 \text{ } 2 \text{ } 239 \text{ } 94 \text{ } + \alpha \rightarrow \text{n} +$  The radionuclide formed is Cm-242. 7. One possible result of the impact of a neutron on a uranium-235 nucleus is the splitting of the uranium into tellurium-137, zirconium-97 and two other particles. Write the nuclear equation for this

### KEY - Nuclear Equations

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### Nuclear Equations Worksheet With Answers

Write a balanced nuclear equation for a natural transmutation. Prerequisites atomic symbols including mass number, atomic number, and charge Information  $4 \text{ He} =$  a helium nucleus, also known as an alpha particle = an electron, also known as a beta particle when emitted by a nucleus Model The following are two nuclear reaction equations:  $4 \text{ He} \dots$

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Nuclear Equation Practice Answer Key Page 8/30. Read PDF Writing Nuclear Equations Answer Key Write the nuclear equation for this reaction and identify the two other particles.  $\text{n} \text{ } 1 \text{ } 0 \text{ } 97 \text{ } 40 \text{ } 137 \text{ } 52 \text{ } 1 \text{ } 0 \text{ } 235 \text{ } 92 \text{ } \text{U} + \rightarrow \text{Te} + \text{Zr} + 2$  The two other particles formed are neutrons. 8. When Writing Nuclear Equations Answer Key

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Nuclear Equation Practice Answer Key Page 8/30. Read PDF Writing Nuclear Equations Answer Key Write the nuclear equation for this reaction and identify the two other particles.  $\text{n} \text{ } 1 \text{ } 0 \text{ } 97 \text{ } 40 \text{ } 137 \text{ } 52 \text{ } 1 \text{ } 0 \text{ } 235 \text{ } 92 \text{ } \text{U} + \rightarrow \text{Te} + \text{Zr} + 2$  The two other particles formed are neutrons. 8. When Page 3/11

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ANSWER KEY Nuclear Decay The following atoms all undergo alpha particle emission. Write the complete nuclear equation.  $210 \text{ } 84 \text{ } \text{Po} \rightarrow$  alpha particle +  $\text{Pb-206}$   $238 \text{ } 92 \text{ } \text{U} \rightarrow$  alpha particle +  $\text{Th-234}$   $238 \text{ } 90 \text{ } \text{Th} \rightarrow$  alpha particle +  $\text{Ra-234}$   $222 \text{ } 86 \text{ } \text{Rn} \rightarrow$  alpha particle +  $\text{Po-218}$  The following atoms all undergo beta decay.

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Nuclear Equation Practice Answer Key - aplikasidapodik.com Nuclear Equation Practice Answer Key Write the nuclear equation for this reaction and identify the two other particles.  $\text{n} \text{ } 1 \text{ } 0 \text{ } 97 \text{ } 40 \text{ } 137 \text{ } 52 \text{ } 1 \text{ } 0 \text{ } 235 \text{ } 92 \text{ } \text{U} + \rightarrow \text{Te} + \text{Zr} + 2$  The two other particles formed are neutrons. 8. When bombarded with neutrons, lithium-6 produced an alpha particle and an

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Write nuclear equations for the following bombardment reactions. HINT: Neutron =  $^0_1n$  Proton =  $^1_1p$  a. Platinum-196 is bombarded by a deuteron ( $^2_1H$ ), producing platinum-197 and a proton.  $^{14}_7N + ^1_0n \rightarrow ^{14}_6C + ^1_1p$  b. Nitrogen-14 is bombarded by a neutron, producing carbon-14 and a proton.  $^{239}_{94}Pu + ^4_2He \rightarrow ^{243}_{96}Cm + 3^1_0n$  c. Plutonium-239 plus an alpha yields three neutrons and a transuranium ...

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Write and balance nuclear equations Changes of nuclei that result in changes in their atomic numbers, mass numbers, or energy states are nuclear reactions . To describe a nuclear reaction, we use an equation that identifies the nuclides involved in the reaction, their mass numbers and atomic numbers, and the other particles involved in the reaction.

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